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answer the following SECTION 7.1 IONS
For each element below, state (i) the number of valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most stable ion. a. Ba 2.

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182 Core Teaching Resources _____ 12.
The molecular structure of carbon dioxide is one carbon atom with two oxygen atoms on opposite sides of it.

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_____ 13. Covalent bonds exist when combining atoms give up or accept electrons. _____ 14. A molecule contains two atoms. Part C Matching Match each description in Column B to the correct term in Column A.

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181 MOLECULAR COMPOUNDS 8**

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answer the following SECTION 71 IONS
For each element below, state (i) the
number of valence electrons in the
atom, (ii) the electron dot structure, and
(iii) the chemical symbol(s) for the most

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