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51. Write a point of distinction between a metallic solid and an ionic solid other than metallic lustre. Answer: Metallic solids can conduct electricity due to movement of electrons in solid state, whereas ionic solids conduct electricity in molten state or in aqueous solution only due to the movement of ions. 52.

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If NaCl is doped with 10-4 mol % of SrCl 2, the concentration of cation vacancies will be (N A = 6.023 x 10 23 mol-1) (a) 6.023 x 10 15 mol-1 (b) 6.023 x 10 16 mol-1 (c) 6.023 x 10 17 mol-1 (d) 6.023 x 10 14 mol-1

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